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SUBJECT CODE NO:- B- 2007 FACULTY OF SCIENCE AND TECHNOLOGY

B.Sc. F.Y. (Sem-II)

Examination November/December- 2022 Chemistry Paper-IV (Physical Chemistry)

Tim	ie: 1:30 Hours] [Max. Mar]	<u>x</u> s:50
	Please check whether you have got the right question paper.	
N.B	 Attempt all questions. Draw diagram wherever necessary. 	, DE
Q.1	Explain postulates of kinetic theory of gases. Deduce Avogadro's law and Graham's law of diffusion. OR	20
	Derive Bragg's equation. Discuss in detail laws of crystallography.	20
Q.2	Explain factors affecting the rate of reaction. Derive integrated rate equation for first order reaction.	20
CARA CARA	Write short note on any four a) Calculate the distance between two points lying on a straight line i) (-4, 3) and (2, -3) ii) (-2, 4) and (5, 4) b) Using logarithms, solve i) 797 ÷ 110 ii) 619 × 321 c) Differentiate between solid and gases d) Nematic and cholesteryl liquid crystals	20
	e) Emulsion and its types f) Enzyme catalysis.	
Q.3	Multiple choice questions.	10
	1) The ideal gas equation PV = a) nT b) nR c) nRT d) None of these	
	2) $V \propto T$ is	

- 3) $k = \frac{1}{ta} \cdot \frac{x}{(a-x)}$ represents ----- order of reaction.
 - a) Zero
- b) First
- c) Pseudo
- d) Second
- 4) The rate of reaction increases with the ----- of temperature.
 - a) Decrease
- b) Increase
- c) Both a & b
- d) None of these

- 5) P-cholesteryl benzoate is an example of ----
 - a) Solid
- b) Liquid
- c) Gases
- d) Liquid crystal

- 6) Ice-cream is an example of ----
 - a) Emulsion
- b) Gel
- c) Aerosol
- d) Sol

- 7) $n\lambda = 2d\sin\theta$ represents ----- law
 - a) Boyle's
- b) Charles's
- c) Bragg's
- d) Both a & b

- 8) HCl is a ----- molecule
 - a) Polar
- b) Non-polar
- c) Both a & b
- d) None of these

- 9) Log 300 = ----
 - a) 1.00
- b) 2.00
- c) 3.00
- d) 4.00
- 10) The slope and intercept of the line x + 2y + 6 = 0 is ----
 - a) $(\frac{1}{2}, 3)$
- b) $\left(-\frac{1}{2}, -3\right)$
- c) $(\frac{-1}{2},3)$
- d) $(\frac{1}{2}, -1)$