

Total No. of Printed Pages:3

**SUBJECT CODE NO:- B-2029**  
**FACULTY OF SCIENCE AND TECHNOLOGY**  
**B.Sc. T.Y (Sem-VI)**  
**Examination November/December- 2022**  
**Physics Paper-XIX**  
**(Atomic, Molecular Physics & Laser)**

[Time: 1:30 Hours]

[Max. Marks: 50]

Please check whether you have got the right question paper.

- N.B
1. All questions are compulsory.
  2. All questions carry equal marks.
- Q. 1
- a) Draw energy level diagram and explain series of spectral line in Bohr's theory of hydrogen atom. 10
  - b) Discuss "Vector atom model" in detail. 10
- OR
- a) What is Raman effect. Explain stokes and antistokes lines in Raman spectrum. 10
  - b) What is LASER? Explain construction and working of CO<sub>2</sub> LASER. 10
- Q. 2
- a) Write a short note on Thomsons atom model. 05
  - b) Write a short note on Rayleigh's law of scattering. 05
  - c) Calculate the radius of electron in Bohr's first orbit of hydrogen atom. Given 05  
 $-e = 1.6 \times 10^{-19} C, m = 9.1 \times 10^{-31} kg, h = 6.625 \times 10^{-34} J.s,$   
 $\epsilon_0 8.85 \times 10^{-12} C^2 / Nm^2$
  - d) Calculate the rotational energy of diatomic molecule corresponding to J=2 level, if the rotational energy corresponding to J=1 level is  $0.7 \times 10^{-32} J$ . 05
- OR
- a) Explain L-S coupling scheme with vector diagram. 05
  - b) What are properties of LASER beam? 05
  - c) Calculate the wavelength separation between component lines of wavelength 6000 A.U. 05  
in normal Zeeman Effect. The magnetic field used is  $0.2 \text{ wb/m}^2$ , specific charge is  $1.76 \times 10^{11} \text{ c/kg}$

- d) Find the ratio of population inversion of the two states in a He-Ne LASER, that produces 05 a light of wavelength 6328 Å. U at 27° C.

Q. 3 Choose the correct answer.

10

- The splitting of the spectral lines under the influence of magnetic Field is \_\_\_\_\_  
 a) Stark effect    b) Photoelectric effect    c) Zeeman effect    d) Compton effect
- The wavelength of Lyman series is obtained in \_\_\_\_\_  
 a) Visible region    c) Ultraviolet region  
 b) Infrared region    d) Far-infrared region
- The value of magnetic spin quantum number are \_\_\_\_\_  
 a)  $\frac{+1}{2}, 0$     b)  $\frac{-1}{2}, 1$     c)  $1, -1$     d)  $\frac{-1}{2}, \frac{+1}{2}$
- The Zeeman effect is a \_\_\_\_\_ phenomenon in which spectral lines are split-up into several components.  
 a) Magnetic    b) Optical    c) Electrical    d) Magneto-optical
- In rotational spectra all frequency lines are \_\_\_\_\_  
 a) Equally spaced    b) Unequally spaced  
 e) *an integral multiple of  $\nu_0$ .*  
 d) *equally spaced and an integral multiple of  $\nu_0$ .*
- Rotational spectra are observed in \_\_\_\_\_  
 a) Visible region    c) Far infrared region  
 b) Microwave region    d) Microwave and far infrared region.
- The process by which atoms are raised from lower level to upper level is called \_\_\_\_\_  
 a) Light amplification    c) Laser pumping  
 b) Population inversion    d) Stimulated emission
- A LASER beam consist of \_\_\_\_\_  
 a) Light material particles    c) Highly coherent photons  
 b) Electrons    d) Cosmic rays

9. If 'r' is radius of first Bohr's orbit then the radius of second Bohr's orbit is \_\_\_\_\_

- a) r                                      b) 2r                                      c) 3r                                      d) 4r

10. If  $L = 1, S = \frac{3}{2}$  then  $j = ?$

- a)  $\frac{3}{2}$                                       b)  $\frac{5}{2}$                                       c)  $\frac{7}{2}$                                       d) 3