SUBJECT CODE NO:-2009

FACULTY OF SCIENCE AND TECHNOLOGY

B.Sc. Third Year Sixth Semester EXAMINATION JUNE/JULY2022

Chemistry Paper –XVI
(Inorganic Chemistry)

[Time:	1.53	Hours]	x.Marks:5
N.B		Please check whether you have got the right question paper. i) Attempt all questions	
		ii) Draw neat and labeled diagrams wherever necessary.	55.77.72 57.77.72
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Q.1	a)	Discuss the splitting of d- orbitals in tetrahedral metal complexes.	10
	b)	Give two methods of preparation, properties and uses of organotitanium compounds. OR	
	c)	Discuss the biological importance of ca++ and Mg++ ions.	10
	d)	67,201.49 (27,201.49) (27,201.49) (27,201.49) (27,201.49) (27,201.49) (27,201.49) (27,201.49)	10
Q.2	a)	What are metal carbonyls? Discuss the nature of bonding in metal carbonyls.	10
	b)	Explain i) Types of Electronic transitions ii) Spin selection rule	10
	c)	Write short notes (any four) 1) $\Delta t = 4/9 \Delta_o$ explain it	20
		 2) Applications of Thin layer chromatography 3) Biological importance of K⁺ ions 4) Spectro chemical series 	
	S	5) Metal ethylenic complexes	
£		6) Orgel energy level diagram of d' system	
Q.3	Multii	ple choice questions	10
2000	~ X , (1), \$	The strong field ligand is	
		a) F^- b) Cl^- c) I^- d) CN^-	
	2)	C.F.S.E of d^3 system in strong field octahedral complex is a) +6Dq b) -4 Dq c) -12 Dq d) +12 Dq	
	3)	The ground term symbol for d^6 system is a) 5_D b) 6_S c) 4_F d) 2D	
	4)	laporte allowed transition is	
A SAN		a) $\Delta l = 0$ b) $\Delta l = \pm 1$ c) $\Delta s = 0$ d) $\Delta s = 1$	

- 5) The central metal ion present in chlorophyll is
 - a) *Fe*⁺⁺
- b) Mg^{++}
- c) Ni^{++} d) Cu^{++}
- 6) The role of hemoglobin in bio system is
- a) Transport of O₂ b) Store O₂ c) store CO₂ d) None of these
- 7) Paper chromatography is an example of
 - a) Adsorption chromatography
 - b) Partition chromatography
 - c) Gas chromatography
 - d) All of these
- 8) In descending paper chromatography the solvent flows in
 - a) Upward direction b) Downward direction c) circular direction d) None of these

- 9) IUPAC name of –(CH₃)₂ Zn is
 - a) Methyl zinc
- b) diethyl zinc
- c) dimethyl zinc d) ethyl methyl zinc
- 10) For polymerizing ethylene to form polythene by

 - a) Organo Al compounds b) Organo Hg compounds
 - c) Organo Zn compounds d) none of these